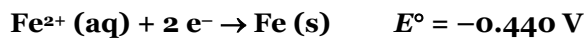


Electrochemistry Quiz

Name: _____

May I post your solution? Yes No Yes, but redact my name

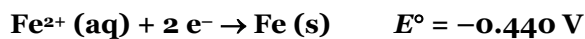
Consider a Galvanic/voltaic cell based on the following half reactions under standard conditions.

What will be the cell potential (E_{cell}) when the cathode solution *changes* by 0.827 M?You may leave your answer in the form of an expression.**Electrochemistry Quiz**

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