Kinetics I Quiz		Name:	
May I post your solution?	[ ] Yes	[ ] No	[ ] Yes, but redact my name
Consider the following rate da	ta for the reaction:	$A + B \rightarrow C$	
Experiment	[A] <sub>0</sub> (M)	[B] <sub>0</sub> (M)	Initial Rate (M/s)
1	0.100	0.200	0.001762
2	0.100	0.400	0.003524
3	0.400	0.200	0.001762

Determine the value of the rate constant (k) for this reaction, including the units.

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Consider the following rate data for the reaction:		$A + B \rightarrow C$	
Experiment	[A] <sub>0</sub> (M)	[B] <sub>0</sub> (M)	Initial Rate (M/s)
1	0.100	0.200	0.001762
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Determine the value of the rate constant (k) for this reaction, including the units.