Kinetics I Quiz

May I post your solution? [ ] Yes
May I post your solution? [ ] Yes
Name:

Consider the following rate data for the reaction:

| Experiment | $[\mathrm{A}]_{\mathrm{o}}(\mathrm{M})$ | $[\mathrm{B}]_{\mathrm{o}}(\mathrm{M})$ | Initial Rate $(\mathrm{M} / \mathrm{s})$ |
| :---: | :---: | :---: | :---: |
| 1 | 0.100 | 0.200 | 0.001762 |
| 2 | 0.100 | 0.400 | 0.003524 |
| 3 | 0.400 | 0.200 | 0.001762 |

Determine the value of the rate constant $(k)$ for this reaction, including the units.

## Kinetics I Quiz

Name:

May I post your solution? [ ] Yes [ ] No [ ] Yes, but redact my name
Consider the following rate data for the reaction: $\quad \mathbf{A}+\mathbf{B} \rightarrow \mathbf{C}$

| Experiment | $[\mathrm{A}]_{\mathrm{O}}(\mathrm{M})$ | $[\mathrm{B}]_{\mathrm{o}}(\mathrm{M})$ | Initial Rate $(\mathrm{M} / \mathrm{s})$ |
| :---: | :---: | :---: | :---: |
| 1 | 0.100 | 0.200 | 0.001762 |
| 2 | 0.100 | 0.400 | 0.003524 |
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Determine the value of the rate constant $(k)$ for this reaction, including the units.

