

Complete each reaction by drawing the correct missing reactant or product.

$$+$$
 Br_2 \longrightarrow Br_2

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$$+ Cl_{2} \rightarrow Cl$$

$$+ Br_{2} \rightarrow Br$$

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$$+$$
 H—OH $\xrightarrow{\text{H}_2\text{SO}_4}$

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+ H-OH
$$H_2SO_4$$
 OH

+ H-OH (no reaction; needs acid)

+ H-OH H^+

or + H-OH H_2SO_4

OH

OH

OH

OH

OH

OH

Which of the following reaction(s) would yield product with a new chiral center?

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$$+$$
 H $-$ OH $\xrightarrow{H^+}$ OH