

What are the unique products for the radical halogenation of pentane with Cl<sub>2</sub> and UV radiation.

$$+ CI - CI \xrightarrow{hv}$$

$$+ CI - CI \xrightarrow{hv}$$

$$+ CI - CI \xrightarrow{hv}$$

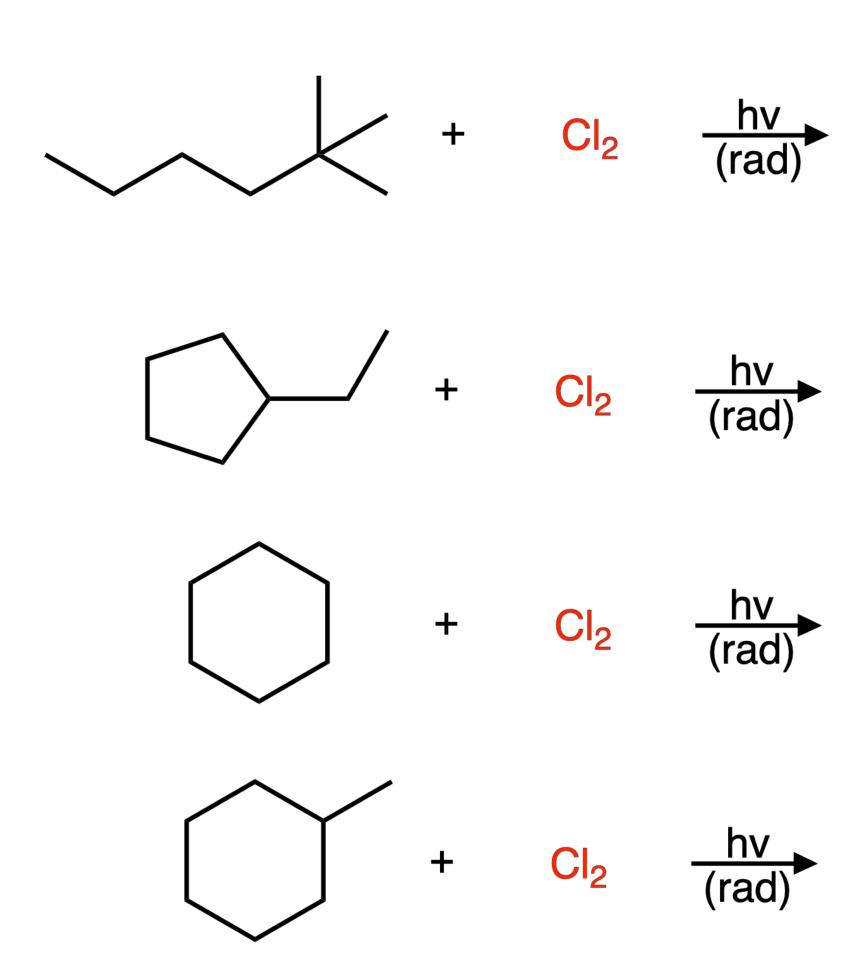
What are the unique products for the radical halogenation of pentane with Cl<sub>2</sub> and UV radiation.

- answer -

Q: Which of the two products of the reaction above contains a chiral center?

A: 2-chloropentane because C2 is chiral.

Give the product(s) for the following alkane radical halogenation reactions.



Give the product(s) for the following alkane radical halogenation reactions.

$$+ Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array} \right\} + Cl_{2} \xrightarrow{hv} \left\{ \begin{array}{c} \\ \\ \\ \end{array}$$

Give the products for the following nucleophilic substitution reactions.

Give the products for the following nucleophilic substitution reactions.

Fill in the red blank squares with the correct reagents or products to make 2-methylhexan-2-ol from 2-methylhexane.



Fill in the red blank squares with the correct reagents or products to make 2-methylhexan-2-ol from 2-methylhexane.

$$+ \boxed{\text{Br}_2} \xrightarrow{\text{hv}} \left( \begin{array}{c} \\ \\ \\ \\ \end{array} \right) + \boxed{\text{\Theta}_{\text{OH}}} \xrightarrow{\text{OH}}$$